- LOEB, A. L. (1958). *Acta Cryst.* 11, 469.
- LOEB, A. L. (1962). *Acta Cryst.* 15, 219.
- MoRRIs, I. L. & LOEB, A. L. (1960). *Acta Cryst.* 13, 434.
- PATTERSON, A. L. & KASPER, J. S. (1959). *International Tables for X-ray Crystallography.* Vol. II, p. 342. Birmingham: Kynoch Press.
- SEITZ, F. (1934). *Z. Kristallogr.* 88, 433.
- SEITZ, F. (1935). *Z. Kristallogr.* 90, 289; 91, 336.
- SEITZ, F. (1936). *Z. Kristallogr.* 94, 100.
- VERWEY, E. J. W. & HEILMANN, E. L. (1947). *J. Chem. Phys.* 15, 174.
- VERWEY, E. J. W., HAAYMAN, P. W. & ROMEIJN, F. C. (1947). *J. Chem. Phys.* 15, 181.
- WAREKOIS, E. P., LAVINE, M. C., MARIANO, A. N. & GATES, H. C. (1962). *J. Appl. Phys.* 33, 690.
- WELLS, A. F. (1962). *Structural Inorganic Chemistry.*  Oxford: Clarendon Press.
- ZHDANOV, G. S. (1945). *Compt. rend. acad. sci. URSS,*  **48,** 39.

*Acta Cryst.* (1964). 17, 1568

# **The Crystal Structures of Nb<sub>3</sub>Se<sub>4</sub> and Nb<sub>3</sub>Te<sub>4</sub>**

### BY KARI SELTE AND ARNE KJEKSHUS

*Kjemisk Institutt A, Universitetet i Osto, Blindern, Norway* 

*(Received* 18 *November* 1963)

The compounds  $Nb_3Se_4$  and  $Nb_3Te_4$  are isostructural. The hexagonal unit cells have the dimensions:

Nb<sub>3</sub>Se<sub>4</sub>: 
$$
a = 10.012
$$
,  $c = 3.4707$  Å;  
Nb<sub>3</sub>Te<sub>4</sub>:  $a = 10.671$ ,  $c = 3.6468$  Å.

The space group is  $P6_3/m$  ( $C_{6h}^2$ ) and the unit cell contains 6Nb and 8Te(Se) in the positions:

6 Nb in  $(h) \pm (x, y, \frac{1}{4}; \overline{y}, x - y, \frac{1}{4}; y - x, \overline{x}, \frac{1}{4})$ with  $x_1 = 0.4886$ ,  $y_1 = 0.1039$ . 2 Te in (c)  $\pm (\frac{2}{3}, \frac{1}{3}, \frac{3}{4}).$ 6Te in (h) with  $x_2 = 0.3389$ ,  $y_2 = 0.2731$ .

The relationships to the  $Ti_sTe_4$  type structure of the  $Nb_5Se_4$  and  $Nb_5Te_4$  phases and the  $NbS_2$  (H)like structure of the  $Nb_{1+x}Se_2$  phase are discussed.

#### **Introduction**

In recent communications by Selte & Kjekshus (1963, 1964) the crystallographic properties of the  $Nb<sub>5</sub>Te<sub>4</sub>$ ,  $Nb<sub>5</sub>Se<sub>4</sub>$  and  $Nb<sub>1+x</sub>Se<sub>2</sub>$  phases have been described. The present paper is concerned with the crystal structures of two previously unknown niobium selenide and telluride\* phases. The compositions were found to be  $Nb_3Se_4$  and  $Nb_3Te_4$  by X-ray and density measurements.

A variety of samples were prepared by heating accurately weighed quantities of the components (in different proportions) in evacuated and sealed silica tubes. The samples were heated at various temperatures between 500 and 1350 °C and quenching in ice water as well as slow cooling was used. Single crystals, grown from the vapour phase, were found in the telluride samples. Certain difficulties in the preparation of the samples resulted from reaction between the niobium and the silica (Selte & Kjekshus, 1963, 1964).

The atomic arrangement in these structures is of considerable interest, especially its resemblance to the neighbouring phases in the niobium-selenium and niobium-tellurium systems, *i.e.* the  $Nb<sub>5</sub>Se<sub>4</sub>, Nb<sub>1+x</sub>Se<sub>2</sub>$ ,  $Nb<sub>5</sub>Te<sub>4</sub>$  and  $NbTe<sub>2</sub>$  phases.

### **Unit cell and space** group

Guinier photographs (taken with strictly monochromatized Cu  $K_{\alpha_1}$  radiation,  $\lambda_{\alpha_1}=1.54050$  A, with potassium chloride,  $a=6.2919~\text{A}$  (Hambling, 1953) added as internal standard) could be indexed on hexagonal axes with the following unit cell dimensions *(cf.* Table 1):

 $Nb_3Se_4$ ;  $a=10.012$ ,  $c=3.4707$  Å,  $c/a=0.3467$ .  $Nb<sub>3</sub>Te<sub>4</sub>; a=10.671, c=3.6468~\AA, c/a=0.3418.$ 

The lattice dimensions were approximately constant

<sup>\*</sup> The niobium tellurides have also been studied by Grigorjan, Simanov & Novoselova (1960) and by Novoselova, Grigorjan & Simanov (1960). They-reported *inter alia* the existence of a phase with composition in the range  $NbTe_{1.00}$ to NbTe<sub>1</sub>.<sub>70</sub>. The available data indicate that their  $NbTe_{1.00-1.70}$ phase is identical with the present  $Nb<sub>3</sub>Te<sub>4</sub>$  phase.

over a range of composition of the starting material for each substance. This shows that the ranges of homogeneity must be rather narrow.

On the basis of the observed densities at  $25.00$  °C, 6.70 g.cm<sup>-3</sup> (Nb<sub>3</sub>Se<sub>4</sub>) and 7.20 g.cm<sup>-3</sup> (Nb<sub>3</sub>Te<sub>4</sub>), the unit cell contains  $2Nb_3X_4$  groups ( $Z_c=2.04$  for Nb<sub>3</sub>Se<sub>4</sub> and  $Z_c = 1.98$  for Nb<sub>3</sub>Te<sub>4</sub>).

The only systematically missing reflexions were of the type

00l when 
$$
l = 2n + 1
$$
.

The Laue symmetry is *6/m* and the possible space groups are  $P6_3$  ( $C_6^6$ ) and  $P6_3/m$  ( $C_{6h}^2$ ).

Table 1. *Guinier photograph data of* Nb<sub>3</sub>Se<sub>4</sub> and Nb<sub>3</sub>Te<sub>4</sub> *taken with strictly monochromatized* Cu  $K_{\alpha_1}$  *radiation* 

| $\sin^2 \theta \times 10^5$ | I <sub>o</sub> |                  |  |
|-----------------------------|----------------|------------------|--|
| $Nb_3Se_4$                  |                | $Nb_3Te_4$       |  |
| S                           | 698            | s                |  |
| vw                          | 2088           | $\boldsymbol{w}$ |  |
| vw                          | 4858           | vw               |  |
| w                           | 5156           | w                |  |
| $\boldsymbol{w}$            | 6258           | $\boldsymbol{m}$ |  |
| vw                          | 6545           | $\boldsymbol{m}$ |  |
| $\boldsymbol{m}$            | 7248           | $\mathit{vs}$    |  |
| $\boldsymbol{m}$            | 9035           | S                |  |
| s                           | 9330           | $\overline{v}$ s |  |
| w                           | 10713          | vw               |  |
| vw                          | 11115          | vw               |  |
| $\boldsymbol{m}$            | 12799          | $\boldsymbol{m}$ |  |
| vw                          | 13192          | vw               |  |
| $\boldsymbol{w}$            | 13486          | $\boldsymbol{w}$ |  |
| w                           | 14592          | $\boldsymbol{w}$ |  |
|                             | 17362          | vw               |  |
| $\boldsymbol{m}$            | 17644          | $\boldsymbol{w}$ |  |
| $\boldsymbol{m}$            | 17854          | w                |  |
| vw                          | 18561          | vw               |  |
|                             | 18721          | vw               |  |
|                             | 19073          | vw               |  |
| w                           | 19437          | $\boldsymbol{w}$ |  |
| vw                          | 21515          | vw               |  |
|                             | 21813          | w                |  |
| vw                          | 23209          | vw               |  |
| vw                          | 24121          | vw               |  |
|                             |                |                  |  |

### Determination of the structure of  $Nb<sub>3</sub>Te<sub>4</sub>$

Well formed, needle-like, single crystals of Nb<sub>3</sub>Te<sub>4</sub> with hexagonal cross-sections were investigated in an integrating Weissenberg camera with Cu K radiation. Intensity measurements of the *hkO* and *hkl*  reflexions were carried out microphotometrically (except for the weakest reflexions, which were estimated visually by comparison with a calibrated scale) on the Weissenberg photographs, using the multiple-film technique. Corrections for the combined Lorentz and polarization factor were made. No corrections for absorption, extinction or temperature factors were applied. For the calculation of  $F_c$  values the atomic scattering factors were taken from Forsyth & Wells (1959) for the *hkO* reflexions and from Vand, Eiland & Pepinsky (1957) for the *hkl* reflexions.

The agreement between  $F_o$  and  $F_c$  is expressed by the reliability index  $R = \sum ||F_o| - |F_c|| / \sum |F_o|$ .  $F_c$  values corresponding to the weakest reflexions, which were determined less accurately than the others, or to undetected reflexions, were not used in the calculation of R indices.

As both space groups have symmetry  $p6$  in the (001) projection, it was decided to start with a Patterson projection on the basis of the corrected  $F<sub>0</sub><sup>2</sup>(hk0)$ values. The [001] direction is specially suitable because of the short c axis and because the crystal has a section not far from circular along the c axis. Tentative atomic positions could be deduced from the Patterson map and a set of structure factors calculated on this basis gave  $R=0.278$ . A Fourier synthesis was evaluated, which clearly showed all atoms resolved. The atomic parameters were refined further by means of two Fourier and two difference syntheses.  $R$  decreased to 0.155 after the Fourier syntheses and to 0.088 after the second difference synthesis. The observed and calculated *F(hkO)* values are listed in Table 2, and the final Fourier map is shown in Fig. 1. The corresponding parameters are:



The short  $c$  axis indicates that the  $z$  parameters for all atoms are 0 and  $\frac{1}{2}$  or  $\frac{1}{4}$  and  $\frac{3}{4}$ .



Fig. 1. Electron-density projection of  $Nb<sub>3</sub>Te<sub>4</sub>$  on (001). Contours are at intervals of  $10 e.\AA^{-2}$ . The zero contours are broken.

The crystals were needle-shaped in the direction of the c axis and the intensities for projections other than (001) would therefore suffer from absorption errors which are difficult to correct for. Intensities of *hkl* reflexions were accordingly evaluated.  $R=0.141$ for  $F(hk1)$  was obtained on the basis of the arrangement suggested in Fig. 2, which in terms of the space group  $P6_3/m$  ( $C_{6h}^2$ ) is as follows:

## Table 2. Observed and calculated structure factors for  $Nb_3Te_4$

The three numbers in each column represent respectively k,  $10F_o$  and  $10F_c$ 





Fig. 2. The structure of  $Nb<sub>3</sub>Te<sub>4</sub>$  viewed along the c axis. Filled circles represent niobium atoms and open circles represent tellurium atoms. The z parameters of the atoms are either  $\frac{1}{4}$  or  $\frac{3}{4}$  as indicated.



The space group and atomic parameters are therefore correctly chosen.

An alternative origin, shifted  $0, 0, \frac{1}{4}$ , is possible and more reasonable for describing the atomic arrangement in this structure. As this origin furthermore is more convenient for bringing out relationships to other structures it is adopted in the rest of this paper.

#### Determination of the structure of  $Nb<sub>3</sub>Se<sub>4</sub>$

Attempts to obtain single crystals of the Nb<sub>3</sub>Se<sub>4</sub> phase were unsuccessful. There are obvious relationships between the intensities of the reflexions from  $Nb_3Se_4$  and  $Nb_3Te_4$  (Table 1) which indicate that the two compounds are isotructural. The rounded off values for the variable parameters  $(x_1=0.49, y_1=0.10,$  $x_2=0.34$  and  $y_2=0.27$  in Nb<sub>3</sub>Te<sub>4</sub> give reasonable agreement between observed and calculated intensities on the Guinier photographs. The parameters for Nb<sub>3</sub>Se<sub>4</sub> cannot be considered very accurate.

#### Discussion of the structures

The coordination around the niobium and selenium (tellurium) atoms can be seen from Fig. 2. The interatomic distances between nearest neighbours are listed in Table 3.

Each niobium atom is surrounded by six selenium (tellurium) atoms at the corners of a deformed octahedron. The structure might be looked upon as built up by these  $NbSe_6(NbTe_6)$  octahedra coupled together by common edges. The niobium atoms are shifted from the centres of the selenium (tellurium) octahedra in a manner which produces alternately longer and shorter niobium-niobium distances. Each Ser atom is surrounded by six niobium atoms at





the corners of a trigonal prism and each  $Se<sub>II</sub>$  atom by four niobium atoms and one hole at the corners of a deformed trigonal bipyramid.

One of the interesting features about these structures is the short niobium-niobium distances, which indicate that bonding exists between the niobium atoms. Niobium-niobium bonds, *i.e.* some short niobium-niobium distances, are predicted by Pauling's (1960) metallic-bond description assuming the usual 'metallic' valences for niobium and selenium (tellurium). However, the observed interatomic distances are not in quantitative agreement with the calculated values based on Pauling's description. The application of the general (8-N) rule *(cf.* Pearson, 1964) to describe transition metal compounds is doubtful when valences and electrical conductivity have not been established. Nevertheless, the rule seems in the present case to



Fig. 3. The structural relationships between the  $Nb_5Se_4$  phase (A: with Ti<sub>5</sub>Te<sub>4</sub> type structure), the Nb<sub>3</sub>Se<sub>4</sub> phase (B) and the  $Nb_{1+x}$ Se<sub>2</sub> phase (C and D: with filled-up NbS<sub>2</sub> (H)-type structure). Filled circles represent niobium atoms and open circles selenium atoms. The numbers indicate fractional heights of the atoms on the projection. Crosses indicate the positions of partially filled niobium lattice sites.



give an almost correct prediction for the number of niobium-niobium bonds. Further comments will have to await more experimental data.

The atomic arrangement in the  $Nb<sub>3</sub>Se<sub>4</sub>$  phase shows interesting similarities to corresponding groups of atoms in the neighbouring  $Nb<sub>5</sub>Se<sub>4</sub>$  and  $Nb<sub>1+x</sub>Se<sub>2</sub>$ phases with respectively Ti<sub>5</sub>Te<sub>4</sub> type and NbS<sub>2</sub>  $(H)$ like structures (Selte & Kjekshus, 1963, 1964). Among the telluride phases the same kinds of relationship exist between the  $Nb<sub>3</sub>Te<sub>4</sub>$  phase and the  $Nb<sub>5</sub>Te<sub>4</sub>$ phase (these phases are isostructural with the corresponding selenide phases) whereas the structure of the NbTe2 phase is not yet known.

In Fig. 3 the structure of  $Nb<sub>5</sub>Se<sub>4</sub>(A)$  is shown in (001) projection, the structure of  $Nb_3Se_4(B)$  in (001) projection and the structure of  $Nb_{1+x}Se_2(C, D)$  in (110) projection for  $x=0.29$ . The axes of projection are of similar dimensions  $(c_{Nb_5Se_4}=3.453$  Å,  $c_{Nb_3Se_4} =$ 3.471 Å and  $a_{\text{Nb}_{1.29}Se_2}=3.450~\text{\AA}$ .

By comparing  $\tilde{A}$  and  $B$  it is seen that the rectangle outlined in B, with edge lengths of  $10.4$  and  $11.2$  Å for the Nb3Se4 phase, has its characteristic counterpart in the square plane cell with  $a=9.871$  Å for the  $Nb_5Se_4$  phase (A). The small dashed quadrilaterals in B resemble those in A. The most notable difference is that the 'central' atoms are selenium atoms in B and niobium atoms in  $A$ . Inside the unit formed by the rectangle in  $B$  there are seven niobium atoms and nine selenium atoms. In order to obtain the  $Nb<sub>3</sub>Se<sub>4</sub>$ structure from that of  $Nb<sub>5</sub>Se<sub>4</sub>$  it is necessary to replace the niobium atoms at the centres of the dashed squares by selenium atoms and to remove one niobium and one selenium atom from the square by a fairly large rearrangement of the surrounding atoms. (This includes twisting and distortion of the dashed squares and movement of one niobium atom from the central dashed square to a 'general' position in the rectangle.)

The structural relationships between the Nb<sub>3</sub>Se<sub>4</sub> phase (B) and the  $Nb_{1+x}Se_2$  phase (C) are rather closer than between the  $Nb<sub>3</sub>Se<sub>4</sub> phase (B)$  and the  $Nb<sub>5</sub>Se<sub>4</sub>$ phase (A). The slightly distorted hexagonal pseudounit indicated in C has edge lengths of 8.8 and 9.0 Å, *i.e.* somewhat smaller than the a axis  $(a=10.012 \text{ Å})$ for  $Nb_3Se_4$ ) in B. These units both contain the same number of atoms, *i.e.* six niobium and eight selenium atoms (including the partly occupied sites in C). The only difference is that one niobium and one selenium atom in each of the two triangles in  $C$  have changed their height by  $\frac{1}{2}$  compared with the similar atoms in B. Within each triangle in B only the selenium atom located at centre of gravity of the triangle has a height differing from those of the rest of the atoms in the same triangle.

A structural relationship between the Nb<sub>5</sub>Se<sub>4</sub> phase and the  $Nb_{1+x}Se_2$  phase is seen by comparing the irregular quadrilateral in  $D$  (with edge lengths of 9.5, 10.3, 10.4 and 10.5 Å) for  $Nb_{1.29}Se_2$  with the square in  $A$ . The smaller dashed quadrilaterals in  $D$ (as previously described for  $B$ ) resemble those in  $A$ except that the central atoms are selenium atoms in D and niobium atoms in A, and that although the four niobium atoms in both cases show an almost square arrangement, two of them have different heights in the structure of the  $Nb_{1+x}Se_2$  phase. Inside the unit formed by the large quadrilateral in  $D$  there are ten selenium atoms and approximately eight niobium atoms when the composition of the phase is  $Nb<sub>1.29</sub>Se<sub>2</sub>$ . In order to obtain the structure of the Nb<sub>5</sub>Se<sub>4</sub> phase it is necessary to substitute the selenium atoms at the centres of the dashed quadrilaterals by niobium atoms and rearrange the vacancies (which occur in the positions indicated by crosses in the structure of the  $Nb_{1+x}Se_2$  phase (D)) so that they are now on the sides of the large quadrilateral.

The structural relationship is much closer between the  $Nb_5Se_4$  phase and the  $Nb_{1+x}Se_2$  phase than between the NbsSe4 phase and the NbsSe4 phase.

No examples are at present known of compounds which are isostructural with  $Nb<sub>3</sub>Se<sub>4</sub>$  and  $Nb<sub>3</sub>Te<sub>4</sub>$ .

The authors express their appreciation to Prof. Haakon Haraldsen for his interest in this study and for making laboratory facilities available.

#### **References**

FORSYTtt, J. B. & WELLS, M. (1959). *Acta Cryst.* 12, 412. GRIGORJAN, L. A., SIMANOV, YU. P. & NOVOSELOVA, A.V.

(1960). *Dokl. Akad. Nauk SSSR,* 135, 1133.

HAMBLING, P. G. (1953). *Acta Cryst.* 6, 98.

- NOVOSELOVA, A. V., GRIGORJAN, L. A. & SIMANOV, YU. P. (1960). *Dokl. Akad. Nauk SSSR,* 135, 864.
- PAULING, L. (1960). The Nature of the Chemical Bond. Ithaca: Cornell Univ. Press.
- PEARSON, W. B. (1964). *Acta Cryst.* 17, 1.
- SELTE, K. & KJEKSHUS, A. (1963). *Acta Chem. Scand.* 17, 2560.
- ~EL~rE, K. & KJ~KSHUS, A. (1964). *Acta Chem. Scand.*  18, 697.
- ~rAND, V., EILAND, P. F. & PEPINSKY, R. (1957). *Acta Cryst.* **10,** 303.